

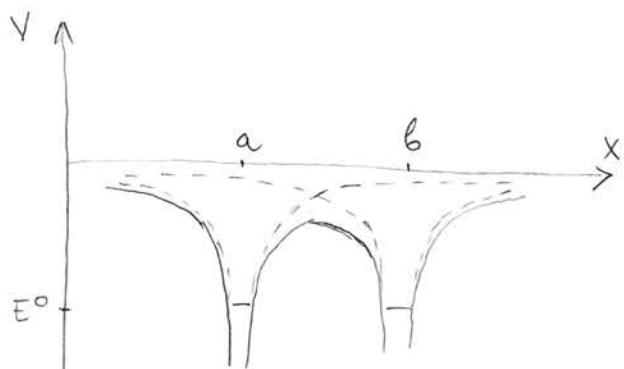
Šred. j-ma koja pričaže ijetanje  $e^-$  u potencijalnom polju jednog a i b imu oblik:

$$\hat{A}\Phi = E\Phi$$

ili.  $\nabla^2\Phi + \frac{2m}{\hbar^2} \left[ E + \frac{e^2}{r_a} + \frac{e^2}{r_b} - \frac{e^2}{R_{ab}} \right] \Phi = 0 \quad \left[ \frac{1}{4\pi\epsilon_0} = 1 \right]$

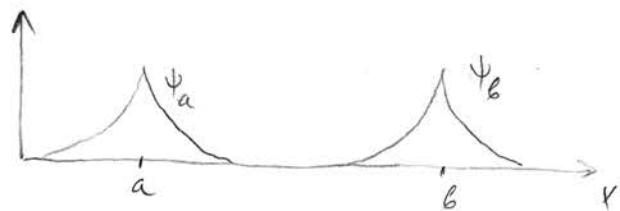
$m$ -masa elektrona;  $\nabla^2$  se odnosi na koordinate elektrona.

Mićemo sada da aproksimativno odredimo  $\Phi$  koristeći method of the linear combination of atomic orbitals (LCAO).



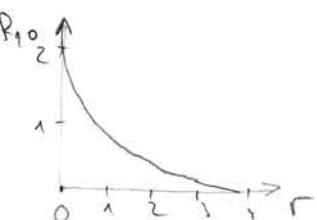
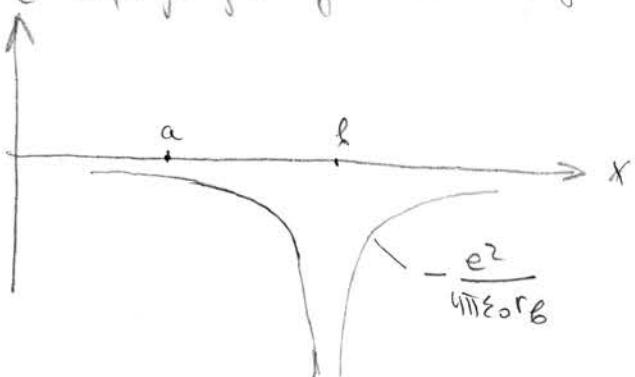
Potenc. energija koja deluje na  $e^-$  usled privlačnih sila od a i b.

Isprekidana kriva pričaže energiju  $e^-$  u polju nukleusa a ili nukleusa b posebno. Prva kriva je isprava potenc. energija. Veživna energija  $e^-$  u polju jednog nukleusa je  $E^0$ .



Goruji grafik: Tal. fja  $\psi_a$  elektrona koji se kreće u polju nukleusa a i odgovarajuća t.f.  $\psi_b$  za  $e^-$  susjednog nukleusa.

Dioni grafik: Ako se rastojanje između a i b smanjuje te počinju t.f.-je da se prepozivaju obično citano



Dakle, pišemo:

$$\Psi = c_a \psi_a + c_b \psi_b$$

$$\psi_a = \frac{1}{\sqrt{\pi a^3}} e^{-r/a} ; \psi_b = \frac{1}{\sqrt{\pi b^3}} e^{-r/b}$$

Odredimo prvo koeficijente  $c_a$  i  $c_b$ .

$$\hat{H} \psi_a c_a + \hat{H} \psi_b c_b = E \psi_a c_a + E \psi_b c_b$$

Izrazimo prvo da leže stvari sa  $\psi_a^*$  pa integralimo, a onda sa  $\psi_b^*$  pa opet integralimo:

$$\left( \int \psi_a^* \hat{H} \psi_a d\tau - E \int \psi_a^* \psi_a d\tau \right) c_a + \left( \int \psi_a^* \hat{H} \psi_b d\tau - E \int \psi_a^* \psi_b d\tau \right) c_b = 0 \quad (1)$$

$$\left( \int \psi_b^* \hat{H} \psi_a d\tau - E \int \psi_b^* \psi_a d\tau \right) c_a + \left( \int \psi_b^* \hat{H} \psi_b d\tau - E \int \psi_b^* \psi_b d\tau \right) c_b = 0 \quad (2)$$

Uvedimo:  $\int \psi_a^* \hat{H} \psi_a d\tau = H_{aa}$  ;  $\int \psi_b^* \hat{H} \psi_b d\tau = H_{bb}$  } Molecular Coulomb integral

$\int \psi_a^* \hat{H} \psi_b d\tau = H_{ab}$  ;  $\int \psi_b^* \hat{H} \psi_a d\tau = H_{ba}$  } Resonance integral.

$\int \psi_a^* \psi_a d\tau = \int \psi_b^* \psi_b d\tau = 1$  ;  $\int \psi_a^* \psi_b d\tau = S_{ab}$  ;  $\int \psi_b^* \psi_a d\tau = S_{ba}$

onda:

$$\begin{vmatrix} H_{aa} - E & H_{ab} - S_{ab} E \\ H_{ba} - S_{ba} E & H_{bb} - E \end{vmatrix} = 0$$

Zbog simetrije  $S_{ab} = S_{ba} = S$

a zbog ermititnosti  $\hat{H}$  biće:  $H_{ab} = H_{ba}$ ;  $H_{aa} = H_{bb}$

$$\Rightarrow \begin{vmatrix} H_{aa} - E & H_{ab} - SE \\ H_{ab} - SE & H_{aa} - E \end{vmatrix} = 0$$

$$(H_{aa} - E)^2 = (H_{ab} - SE)^2 \text{ imamo 2 rješenja}$$

$$H_{aa}^2 - 2E H_{aa} + E^2 = H_{ab}^2 - 2H_{ab} SE + S^2 E^2$$

$$E^2(1-S^2) - 2E(H_{aa} - H_{ab}S) + H_{aa}^2 - H_{ab}^2$$

$$E_{1,2} = \frac{2(H_{aa} - H_{ab}S) \pm \sqrt{(H_{aa}^2 - 2H_{aa}H_{ab}S + H_{ab}^2S^2) - 4(1-S^2)(H_{aa}^2 - H_{ab}^2)}}{2(1-S^2)}$$

$$= \frac{2(H_{aa} - H_{ab}S) \pm \sqrt{4H_{aa}^2 - 8H_{aa}H_{ab}S + 4H_{ab}^2S^2 - 4H_{aa}^2 + 4H_{ab}^2 + 4S^2H_{aa}^2 - 4S^2H_{ab}^2}}{2(1-S^2)} = \frac{2(H_{aa} - H_{ab}S) \pm \sqrt{4(SH_{aa} - H_{ab})^2}}{2(1-S^2)}$$

$$E_{1,2} = \frac{2(H_{aa} - H_{ab}s) \pm 2(sH_{aa} - H_{ab})}{2(1-s^2)} = \frac{H_{aa} - H_{ab}s \pm sH_{aa} - H_{ab}}{1-s^2}$$

$$E_{1,2} = \frac{H_{aa}(1 \pm s) \mp H_{ab}(1 \pm s)}{1-s^2} = \frac{(1 \pm s)(H_{aa} \mp H_{ab})}{(1+s)(1-s)} \Rightarrow \begin{cases} \frac{H_{aa} - H_{ab}}{1-s} = E_A \\ \frac{H_{aa} + H_{ab}}{1+s} = E_S \end{cases}$$

Zároveň  $E_S$  u (1) tj. u  $(H_{aa} - E)c_a + (H_{ab} - sE)c_b = 0$

$$\text{dokážeme: } \left(H_{aa} - \frac{H_{aa} + H_{ab}}{1+s}\right)c_a + \left(H_{ab} - s\frac{H_{aa} + H_{ab}}{1+s}\right)c_b = 0$$

$$\cancel{\left(\frac{(1+s)H_{aa} - H_{aa} - H_{ab}}{1+s}\right)}c_a + \cancel{\left(H_{ab}(1+s) - sH_{aa} - sH_{ab}\right)} \frac{1}{1+s}c_b = 0$$

$$(sH_{aa} - H_{ab})c_a + (H_{ab} - sH_{aa})c_b = 0 \Rightarrow \boxed{c_a = c_b}$$

Zároveň  $E_A$  u (2) tj. u  $(H_{ba} - Es)c_a + (H_{bb} - E)c_b = 0$

$$\left(H_{ba} - \frac{H_{aa} - H_{ab}}{1-s}s\right)c_a + \left(H_{bb} - \frac{H_{aa} - H_{ab}}{1-s}\right)c_b = 0$$

$$[H_{ab}(1-s) - H_{aa}s + H_{ab}s]c_a + [H_{bb}(1-s) - H_{aa} + H_{ab}]c_b = 0$$

$$(H_{ab} - H_{aa}s)c_a + (-H_{aa}s + H_{ab})c_b = 0$$

$$(H_{ab} - H_{aa}s)(c_a + c_b) = 0$$

$$\Rightarrow \text{Imeňo dve rešenia } c_a = c_b \quad \text{i} \quad c_a = -c_b$$

Ove konstanty se odredují z ustanovení normity

$$\int \Phi^* \Phi d\tau = \int (c_a \psi_a^* + c_b \psi_b^*)(c_a \psi_a + c_b \psi_b) d\tau = 1$$

$$\Rightarrow c_a^2 + 2c_a c_b s + c_b^2 = 1$$

$$\text{za slúčj. } c_a = c_b \Rightarrow c_a = c_b = \frac{1}{\sqrt{2(1+s)}}$$

$$\text{i} \quad c_a = -c_b = \frac{1}{\sqrt{2(1-s)}}$$

Dosicle, simetrica f-ja je!

$$\Psi_s = \frac{1}{\sqrt{2(1+s)}} (\psi_a + \psi_b)$$

$$\Psi_A = \frac{1}{\sqrt{2(1-s)}} (\psi_a - \psi_b)$$

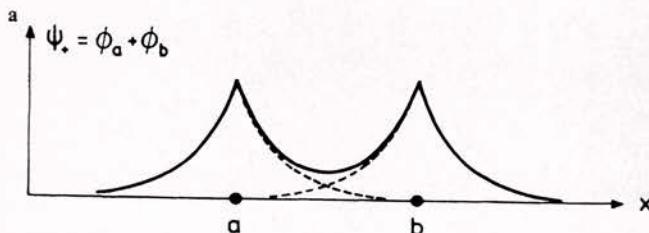


Fig. 24.6. a) The symmetric wavefunction  $\psi_+$  is formed by superposing the wavefunctions  $\phi_a$  and  $\phi_b$ . Because of the overlap between  $\phi_a$  and  $\phi_b$ , the occupation probability for  $\psi_+$  between the two nuclei is increased. b) Illustration of the density distribution of the electron in the  $\psi_+$  state

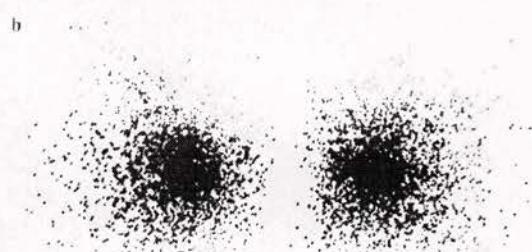
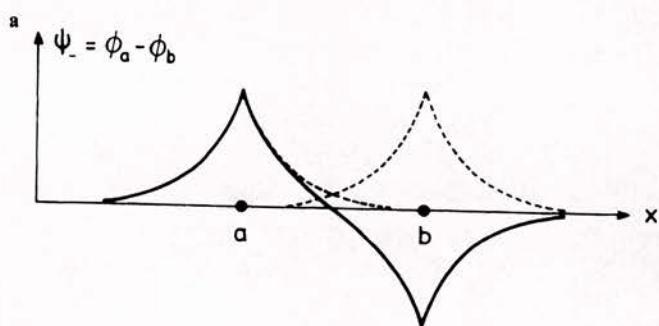
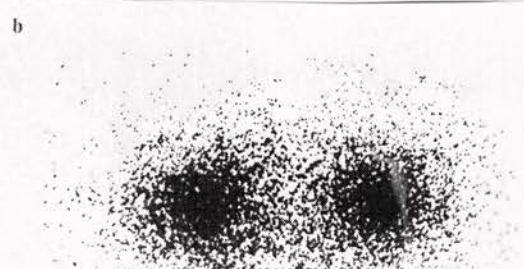


Fig. 24.7. a) The antisymmetric wavefunction  $\psi_-$  is formed by taking the difference of  $\phi_a$  and  $\phi_b$ . The occupation probability is clearly zero in the plane of symmetry between the two nuclei. b) Illustration of the density distribution of the electron in the hydrogen-molecule ion in the case of the antisymmetric wavefunc-

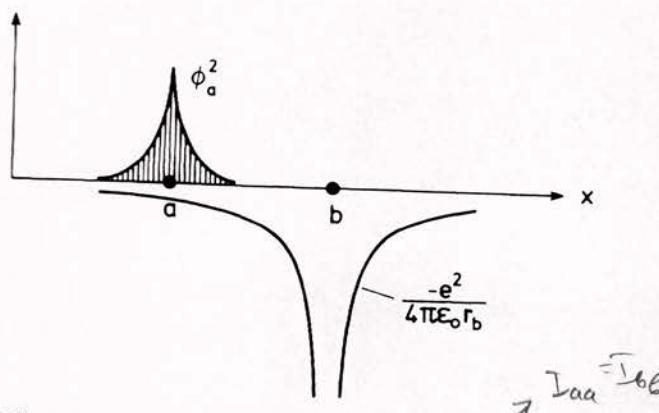


Fig. 24.4. Intuitive explanation of the integral (24.8). This integral represents the Coulomb interaction energy of an electron density cloud with the probability density distribution  $\phi_a^2$  in the Coulomb field of the nucleus. The density distribution is shown as a shaded area, while the full curve gives the potential energy of a point charge in the Coulomb field of nucleus b. To calculate the integral, at each point in space the value of  $\phi_a^2$  must be multiplied with the corresponding value of  $-e^2/(4\pi\epsilon_0 r_b)$  and the products must be added (integrated) over the entire volume

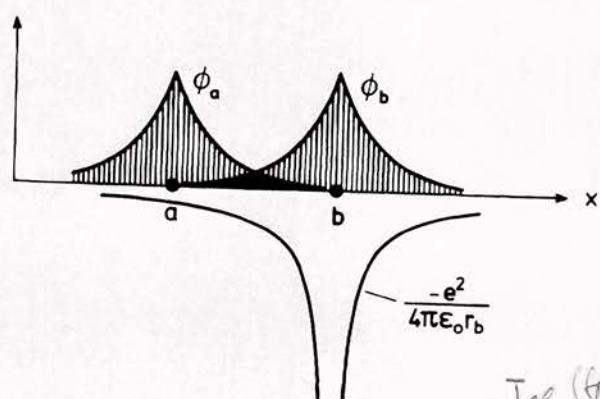


Fig. 24.5. Intuitive explanation of the integral (24.9). The three functions  $\phi_a$ ,  $\phi_b$ , and  $-e^2/(4\pi\epsilon_0 r_b)$  which occur in the integral are plotted. Since the product of these three functions occurs, nonvanishing contributions are only possible when the wavefunctions  $\phi_a$  and  $\phi_b$  overlap, as is indicated by the heavily shaded area. The integral is obtained by multiplying the functional values of  $\phi_a$ ,  $\phi_b$ , and  $-e^2/(4\pi\epsilon_0 r_b)$  at each point in space and adding up over the whole volume

Iab (from p. 16)

14a

## Eliptičke koordinate:

Uzme li se koordinati početak negde u nekoj tački na  $r_{ab}$  i  $z$ -osa duž  $r_{ab} (= R)$  onda se tri eliptičke koordinate tačke u kojoj se nalazi elektron  $(\xi, \eta, \varphi)$  definisu ovako:

$$\xi = \frac{r_a + r_b}{R}; \quad \eta = \frac{r_a - r_b}{R}, \quad \varphi$$

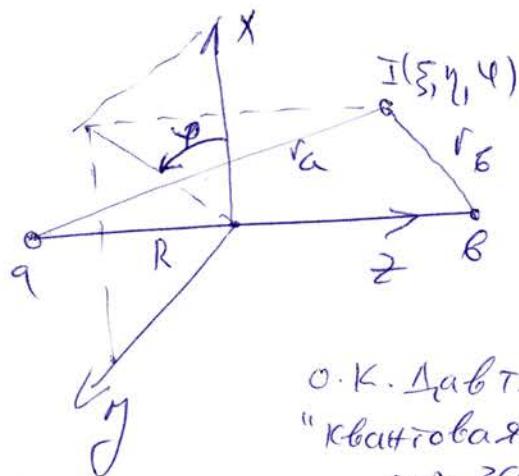
$$\xi \in [1, \infty]; \quad \eta \in [-1, +1]; \quad \varphi \in [0, 2\pi].$$

\*-ugao obrtanja oko preve koga  
spaja jezgra sa  $b$

Inverzne rel. su:

$$r_a = \frac{R}{2}(\xi + \eta); \quad r_b = \frac{R}{2}(\xi - \eta)$$

$$d\tilde{\tau} = \frac{R^3}{8} (\xi^2 - \eta^2) d\xi d\eta d\varphi = dx dy dz$$



О.К. ДАВТЯН  
"Квантовая химија"  
суп. 365:

H<sub>aa</sub> i H<sub>ab</sub> integroli

$$\begin{aligned}
 H_{aa} &= \int \psi_a^* \hat{H} \psi_a d\tau = \int \psi_a^* \left[ -\frac{\hbar^2}{2m} \nabla^2 - \frac{e^2}{r_a} - \frac{e^2}{r_b} + \frac{e^2}{r_{ab}} \right] \psi_a d\tau \\
 &= E_{1s} \int \psi_a^* \psi_a d\tau - e^2 \int \psi_a^* \frac{1}{r_b} \psi_a d\tau + e^2 \int \psi_a^* \frac{1}{r_{ab}} \psi_a d\tau \rightarrow \text{smatruju se } r_a \text{ fixirana je} \\
 \Rightarrow H_{aa} &= E_{1s} - e^2 \langle \psi_a | \frac{1}{r_b} | \psi_a \rangle + \frac{e^2}{R} \\
 &\quad [E_{1s} = E_1 = -13,6 \text{ eV} = -0,15 \text{ a.u.}]
 \end{aligned}$$

Residualni integral:

$$\begin{aligned}
 H_{ab} &= \langle \psi_a | \hat{H} | \psi_b \rangle = \langle \psi_a | \left[ -\frac{\hbar^2}{2m} \nabla^2 - \frac{e^2}{r_a} - \frac{e^2}{r_b} + \frac{e^2}{r_{ab}} \right] | \psi_b \rangle \\
 &= E_{1s} \underbrace{\langle \psi_a | \psi_b \rangle}_S - e^2 \underbrace{\langle \psi_a | \frac{1}{r_a} | \psi_b \rangle}_{S'} + \frac{e^2}{r_{ab}} \underbrace{\langle \psi_a | \psi_b \rangle}_S \\
 H_{ab} &= \left( E_{1s} + \frac{e^2}{R} \right) S - e^2 \langle \psi_a | \frac{1}{r_a} | \psi_b \rangle
 \end{aligned}$$

Za uveličenje preostalih integrala S,  $I_{aa} = \langle \psi_a | \frac{e^2}{r_b} | \psi_a \rangle = I_{bb} = \langle \psi_b | \frac{e^2}{r_a} | \psi_b \rangle$   
 i  $I_{ab} = \langle \psi_a | \frac{e^2}{r_a} | \psi_b \rangle$  koriste se eliptičke koordinate.

$$\xi = \frac{r_a + r_b}{R} ; \quad \eta = \frac{r_a - r_b}{R} ; \quad \varphi$$

$$\xi \in [1, \infty] ; \quad \eta \in [-1, +1] ; \quad \varphi \in [0, 2\pi]$$

$$\text{Inverzne su: } r_a = \frac{R}{2}(\xi + \eta) ; \quad r_b = \frac{R}{2}(\xi - \eta)$$

$$d\tau = \frac{R^3}{8} (\xi^2 - \eta^2) d\xi d\eta d\varphi ;$$

$$\psi_a \psi_b = \frac{1}{\pi a^3} e^{-(r_a + r_b)/a} = \frac{1}{\pi a^3} e^{-R\xi/a}$$

$-e|\psi_a|^2 \rightarrow$  the charge density of the electron

$-e\psi_a \psi_b \rightarrow$  exchange density; this means that electron in some sense is partly in state a and partly in state b, or, in other words, an exchange between these two states takes place.

Integrali koje sadrže  $\psi_a \psi_b$  nisuve se izračunici i oni reprezentuju poseban kvantitativni referat.

### Rešavanje dvocentričnih integrala

$$J = \int \frac{\exp(-pr_a) \exp(-qr_b)}{r_a r_b} d^3 \vec{r}$$

u eliptičnim koordinatama  $J$  postaje

$$J = \frac{R}{2} \int_1^{\infty} d\xi \int_{-1}^{+1} dy \int_0^{2\pi} d\phi \exp(-a\xi - b y)$$

$$\text{gde je } a = \frac{p}{2}(p+q), \quad b = \frac{p}{2}(p-q)$$

$$\Rightarrow J = \frac{\pi R}{ab} e^{-a} (e^b - e^{-b}) = \frac{4\pi}{R} \frac{1}{p^2 - q^2} (e^{-qR} - e^{qR})$$

Drugi relevantni integrali dobijaju se diferenciranjem:

$$K = \int \frac{\exp(-pr_a) \exp(-qr_b)}{r_a} d^3 \vec{r} = -\frac{\partial}{\partial q} J$$

$$K = \frac{4\pi}{R} \left[ \frac{R}{p^2 - q^2} e^{-qR} + \frac{2q}{(p^2 - q^2)^2} (e^{-qR} - e^{qR}) \right]$$

$$L = \int \exp(-pr_a) \exp(-qr_b) d^3 \vec{r} = -\frac{\partial}{\partial p} K$$

$$L = \frac{8\pi}{R(p^2 - q^2)^2} \left[ R(p e^{-qR} + q e^{-pR}) + \frac{4pq}{p^2 - q^2} (e^{-pR} - e^{-qR}) \right]$$

Kada je  $p = q$

$$J = \frac{4\pi}{R} \lim_{q \rightarrow p} \frac{e^{-qR} - e^{-pR}}{p^2 - q^2} = \frac{4\pi}{12} R \frac{e^{-pR}}{2p} = \frac{2\pi}{p} e^{-pR}$$

$$K = \frac{4\pi}{R} \lim_{q \rightarrow p} \frac{R(p^2 - q^2) e^{-qR} + 2q(e^{-pR} - e^{-qR})}{(p^2 - q^2)^2}$$

$$K = \frac{4\pi}{R} \lim_{q \rightarrow p} \frac{R(2pe^{-qR} - 2qe^{-pR})}{2(p^2 - q^2) 2p} = 2\pi \lim_{q \rightarrow p} \frac{pe^{-qR} - qe^{-pR}}{p(p^2 - q^2)}$$

$$K = 2\pi \lim_{q \rightarrow p} \frac{e^{-qR} + Rq e^{-pR}}{p^2 - q^2 + p \cdot 2p} = \frac{\pi}{p^2} (1 + pr) e^{-pR}$$

Slično:

$$L = \frac{\pi}{p^3} \left( 1 + pr + \frac{1}{3} p^2 R^2 \right) e^{-pR}$$

Koristeci rešenja integrala na str. 15 imamo:

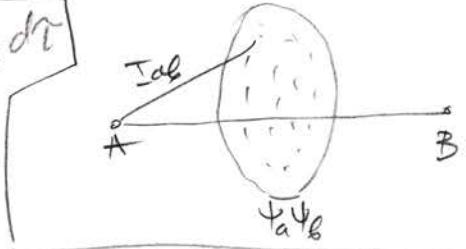
$$S = \langle \psi_a | \psi_b \rangle = \left\{ 1 + \frac{ZB}{a} + \frac{1}{3} \left( \frac{ZB}{a} \right)^2 \right\} e^{-ZB/a} \quad (2=1)$$

I<sub>ab</sub> - predstavlja energiju razmene ili energiju rezonancije. Može i označiti se kao interakcija (interaction) između dve molekule.

$$I_{ab} = \frac{e^2}{a} \left\{ 1 + \frac{ZB}{a} \right\} e^{-ZB/a} \quad (=2)$$

$$I_{aa} = e^2 \int \psi_a \left( \frac{1}{r_B} \right) \psi_a dr = I_{bb} = -e^2 \int \psi_b \left( \frac{1}{r_a} \right) \psi_b dr$$

$$= \frac{e^2}{R} \left[ 1 - e^{-2Z/a} \left( 1 + \frac{ZB}{a} \right) \right] \quad (=3)$$



Sada možemo da pismemo simetrični i antisimetrični

izrat za energiju

$$E_S \equiv E_+ = \frac{H_{aa} + H_{ab}}{1+S} = E_H + \frac{e^2}{R} - \frac{I_{aa} + I_{ab}}{1+S}$$

$$E_A \equiv E_- = \frac{H_{aa} - H_{ab}}{1+S} = E_H + \frac{e^2}{R} - \frac{I_{aa} - I_{ab}}{1-S}$$

Oba integrala I<sub>aa</sub> i I<sub>ab</sub> su pozitivna, sa I<sub>aa</sub> > I<sub>ab</sub>.

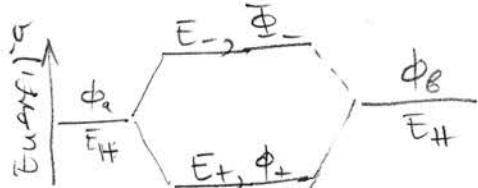
Oba integrala I<sub>aa</sub> i I<sub>ab</sub> su pozitivna, sa I<sub>aa</sub> > I<sub>ab</sub>.

Niža energija je E<sub>S</sub> ( $E_+$ ) i ta orbitala je  $\Phi_S$  ( $\Phi_+$ )

Niža energija je E<sub>A</sub> ( $E_-$ ) i ta orbitala je  $\Phi_A$  ( $\Phi_-$ )

i natinja se veziva orbitala, formira je antibonding orbitala.

Simetrično stanje odgovara stabilitetu molekula



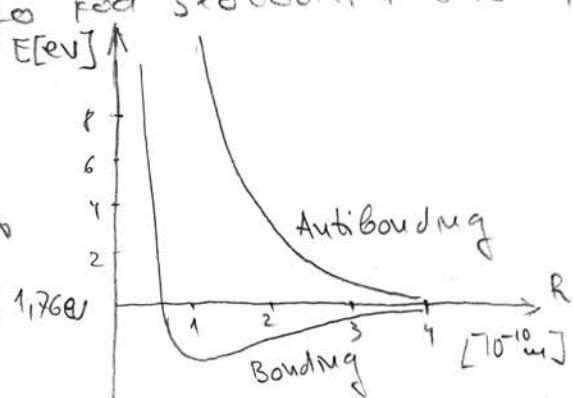
Ne zaboravimo: vezivanje u molekul nastaje kod elektron i dolaze u energetskim stanjima koja su niža nego kod slobodnih atoma.

Energija disocijacije je razlika između

minimalne energije u ravnotežnom

stanju i energije na distanci R = r<sub>ab</sub> → ∞

ovakva teorija daje minimum na R ≈ 1,32 Å i 1,76 eV.  
a ekspl. vrijednosti su 1,06 Å i 2,79 eV.



Laplacijan u eliptičkim koordinatama (confocal elliptic coordinates):

$$\Delta = \nabla^2 = \frac{1}{R^2(\xi^2 - \eta^2)} \left\{ \frac{\partial}{\partial \xi} \left[ (\xi^2 - 1) \frac{\partial}{\partial \xi} \right] + \frac{\partial}{\partial \eta} \left[ (1 - \eta^2) \frac{\partial}{\partial \eta} \right] + \frac{\xi^2 - \eta^2}{(\xi^2 - 1)(1 - \eta^2)} \frac{\partial^2}{\partial \eta^2} \right\}$$

Sr. j-va  $\left( -\frac{1}{2} \Delta - \frac{1}{r_a} - \frac{1}{r_b} + \frac{1}{R} - \epsilon \right) \Phi = 0$

$$\frac{\partial}{\partial \xi} \left[ (\xi^2 - 1) \frac{\partial \Phi}{\partial \xi} \right] + \frac{\partial}{\partial \eta} \left[ (1 - \eta^2) \frac{\partial \Phi}{\partial \eta} \right] + \left( \frac{1}{\xi^2 - 1} + \frac{1}{1 - \eta^2} \right) \frac{\partial^2 \Phi}{\partial \eta^2} + 2R^2 \left[ \frac{1}{4} \left( \epsilon - \frac{1}{R} \right) / (\xi^2 \eta^2) + \frac{1}{R^2} \right] \Phi = 0$$

Rešenje tražimo u vidu pravzoda:

$$\Phi(\xi, \eta, \epsilon) = F(\xi) G(\eta) e^{i\mu\epsilon}, \quad \mu = 0, \pm 1, \pm 2, \dots$$

gde su f-je  $F(\xi)$  i  $G(\eta)$  su normalizabilna rešenja j-va:  $\frac{d}{d\xi} \left[ (\xi^2 - 1) \frac{dF}{d\xi} \right] + \left[ \frac{R^2}{2} \left( \epsilon - \frac{1}{R} \right) \xi^2 + 2R\xi - \frac{\mu^2}{\xi^2 - 1} + \mu \right] F(\xi) = 0 \dots (1)$

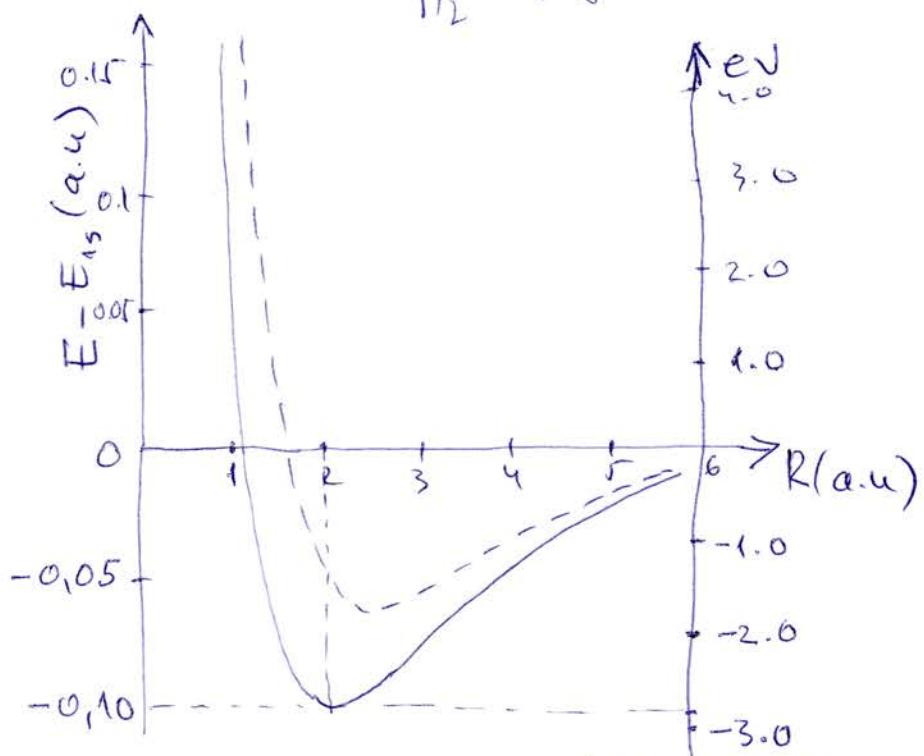
$$\frac{d}{d\eta} \left[ (1 - \eta^2) \frac{dG}{d\eta} \right] - \left[ \frac{R^2}{2} \left( \epsilon - \frac{1}{R} \right) \eta^2 + \frac{\mu^2}{1 - \eta^2} + \mu \right] G(\eta) = 0 \dots (2)$$

$\mu$ -separaciona konstanta. Snovi elektronski fermi je ocaracterisan sa tri kv. broja  $\lambda = |\mu|$ ,  $n_x, n_y$  što predstavlju mreže f-ja  $F(\xi)$  i  $G(\eta)$ .

j-va (1) i (2) mogu da se reše numerički za osnovno i pobudnu stanju. To su prvi uradili 1953. D. Bates, K. Ledsham i A. Stewart

(166)

$H_2^+ (5g\ 1s)$  lowest  $5g$  state

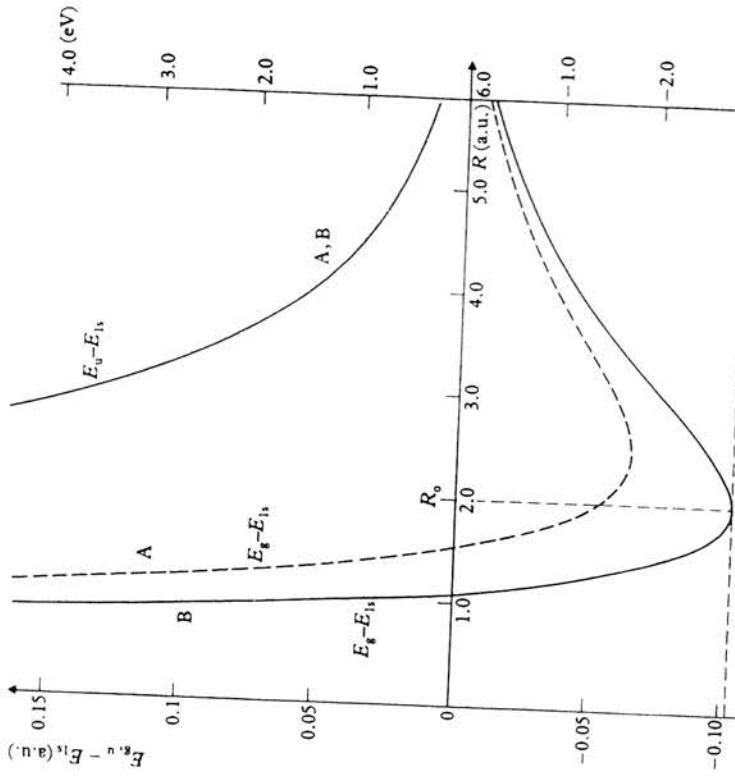


— LCAO  $\rightarrow \begin{cases} \text{Energija dissociacije} \\ D_e = E_{1s} - E_g(R_0) = 0,065 \text{ a.u.} = 1,77 \text{ eV} \\ (R_0 = 2,49 \text{ a.u.} = 1,32 \text{ \AA}) \end{cases}$

— exact values  $(E \equiv E_g)$

$\hookrightarrow \begin{cases} D_e = E_{1s} - E_g(R_0) = 0,103 \text{ a.u.} = 2,79 \text{ eV} \\ R_0 = 1,06 \text{ \AA} \end{cases}$

$1_{a_0} = 0,529 \text{ \AA}$



9.9 The lowest electronic potential energy curves of  $\text{H}_2^+$ . The dashed lines labelled A, show  $(E_s(R) - E_{1s})$  and  $(E_u(R) - E_{1s})$  calculated using the simple LCAO wave functions of [9.47], while the solid curves labelled B show the exact values of the same quantities calculated from equations [9.61] to [9.64].

an attraction leading to the formation of a stable molecular ion. The corresponding molecular orbital  $\Phi_g$  given by [9.47a] is said to be a *bonding* molecular orbital. In the present case it is an approximation to the ground state (that is, the lowest  $\sigma_g$  state) and is designated  $\sigma_g\text{1s}$ .

In contrast, we see from Fig. 9.9 that the function  $E_u(R)$  has no minimum and is repulsive at all distances; a molecule in this state will immediately dissociate into a proton and a hydrogen atom in the 1s state. The corresponding molecular orbital  $\Phi_u$  (see [9.47b]) is called an *antibonding* orbital, and is denoted in the present case by  $\sigma_u^*\text{1s}$ , where the superscript (\*) indicates that we are dealing with an antibonding orbital.

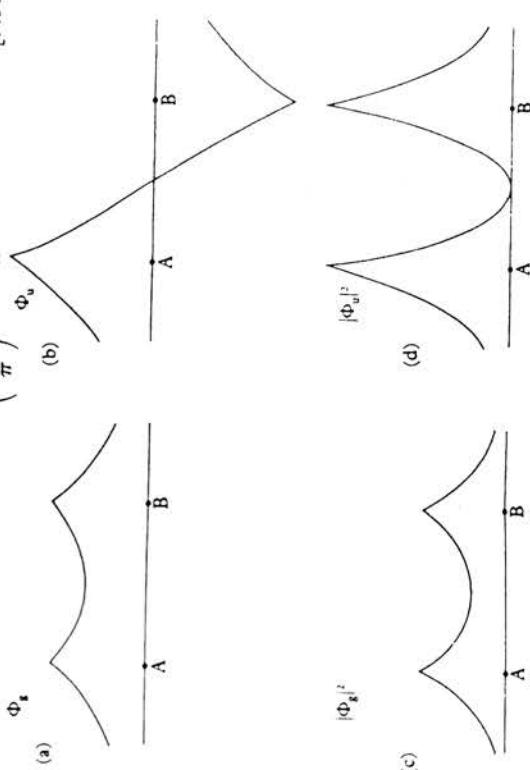
It is worth noting that the term  $H_{AB}$ , in which the matrix element of the Hamiltonian  $H$  between atomic orbitals centred on the protons A and B is evaluated, plays a vital role in obtaining bonding for the  $\sigma_g\text{1s}$  state. If this term

were ignored, the two functions  $E_g(R)$  and  $E_u(R)$  would coincide, the resulting curve being repulsive at all distances.

The electron probability density in the states  $\Phi_g$  and  $\Phi_u$  is given by  $|\Phi_g|^2$  and  $|\Phi_u|^2$ , respectively, so that the corresponding charge densities are  $\rho_g = -e|\Phi_g|^2$  and  $\rho_u = -e|\Phi_u|^2$  (or  $\rho_g = -|\Phi_g|^2$  and  $\rho_u = -|\Phi_u|^2$  in atomic units). If the charge density  $\rho_g$  is evaluated at points between A and B along the internuclear line, it is found to be greater (in absolute value) than the sum of the densities due to two isolated H atoms with their protons placed at A and B, normalised so that half an electron is associated with each proton. It is this *excess of negative charge* between the protons which causes the binding (or bonding). On the other hand, if the charge density  $\rho_u$  corresponding to the antibonding case is evaluated, a *deficiency of negative charge* is found between the protons. This is clearly seen in Fig. 9.10 where the wave functions  $\Phi_g$  and  $\Phi_u$  as well as the absolute value of the charge densities  $\rho_g$  and  $\rho_u$  are plotted along the internuclear line.

The exact binding energy of  $\text{H}_2^+$  is a little greater than the result obtained above, with  $D_e = E_{1s} - E_g(R_0) = 0.103$  a.u. = 2.79 eV, and the true equilibrium distance is  $R_0 = 1.06 \text{ \AA}$ . The principal failing of the approximate wave function  $\Phi_g(\mathbf{R}; \mathbf{r})$  given by [9.47a] is that at small separations  $\Phi_g$  should approach the wave function of  $\text{He}^+(1s)$ , the ground state of the positive helium ion with nuclear charge  $Z = 2$ , and in the approximation [9.47a] it does not. This defect can be remedied by using orbitals of variable charge, such as

$$u_{1s}(Z^*, r) = \left( \frac{Z^{*3}}{\pi} \right)^{1/2} e^{-Z^* r} \quad [9.58]$$



9.10 Wave functions  $\Phi_g$  and  $\Phi_u$  and charge densities  $|\Phi_g|^2$ ,  $|\Phi_u|^2$  for the hydrogen molecular ion  $\text{H}_2^+$ , plotted along the internuclear line to an arbitrary scale. The points A and B represent the positions of the two protons.